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APPLICATION NO.	APPLICATION NO. FILING DATE		FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/535,070	10/535,070 05/12/2005		Johannes Engelbertus Adrianus Maria Van Den Meerakker	NL02 1185 US 5660	
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1109 MCKA			ART UNIT	PAPER NUMBER	
SAN JOSE,	CA 951	31	1765	·	

DATE MAILED: 04/04/2006

Please find below and/or attached an Office communication concerning this application or proceeding.

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		Applicati	on No.	Applicant(s)	W			
Office Action Summary			70	VAN DEN MEERAKKER ET AL.				
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Status								
2a) <u></u>	Responsive to communication(s) filed on 12 This action is FINAL . 2b) To Since this application is in condition for allocation accordance with the practice under	his action is r wance except	for formal matters, pr		e merits is			
Dispositi	on of Claims							
5)	Claim(s) <u>1-13</u> is/are pending in the applicat 4a) Of the above claim(s) is/are with Claim(s) is/are allowed. Claim(s) <u>1-13</u> is/are rejected. Claim(s) is/are objected to. Claim(s) are subject to restriction an	drawn from co						
Applicati	on Papers							
10)	The specification is objected to by the Examember The drawing(s) filed on is/are: a) and a Applicant may not request that any objection to Replacement drawing sheet(s) including the coronath or declaration is objected to by the	accepted or b the drawing(s) rection is requi	be held in abeyance. Se red if the drawing(s) is of	ee 37 CFR 1.85(a). bjected to. See 37 CI				
Priority u	ınder 35 U.S.C. § 119							
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 								
2) Notice	t(s) le of References Cited (PTO-892) le of Draftsperson's Patent Drawing Review (PTO-948) mation Disclosure Statement(s) (PTO-1449 or PTO/SB r No(s)/Mail Date 05/12/05.		4) Interview Summar Paper No(s)/Mail D 5) Notice of Informal 6) Other:	Date	O-152)			

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DETAILED ACTION

Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 3. Claim 1 is rejected under 35 U.S.C. 103(a) as being unpatentable over Van Den Meerakker et al. (Journal of The Electrochemical Society, 147 (7) pages 2757-2761 (2000)) in view of Grüning et al. (US 5,987,208), and further in view of Bielgelsen et al. (US 5,607,876).

The reference of Van Den Meerakker describes a method for etching of deep macropores which will also result in the formation of nanowires if the etching is carried out such as the diameter of the pores, arranged in a hexagonal array (page 2757,

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paragraph "Experimental"), is large enough to allow intersection of the pores leaving narrow regions between pores which will become the "nanowires" formed on the semiconductor material (silicon wafer). The method comprises;

providing a patterned etching mask (silicon nitride) having openings on a surface of a substrate made of the semiconductor material, which openings have a substantially uniform pitch (hexagonal array), placing the substrate with the etching mask in a liquid etchant for the semiconductor material (figure 1); anodically etching so as to form substantially parallel pores with a pitch corresponding to the pitch of the openings in the etching mask (figure 2).

A difference is noted between applicants claim 1 and the reference of Van Den Meerakker cited above, Van Den Meerakker is silent about the diameter of the pores becomes at least as great as the pitch of the pores, whereby nanowires are formed, however, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the method of Van Den Meerakker by widening the diameter of the pores until nanowires are formed because the hexagonal pattern of the pores is conducive to formation of pores and islands (wires) when the pore diameters intersect. One of ordinary skill in the art would have been motivated to form nanowires because nanowires are useful for electro-optical and electromechanical devices.

A second difference is noted between applicants claim 1 and the reference of Van Den Meerakker cited above, Van Den Meerakker is silent about the anodic etching is carried out in a first time period and a second time period, which periods correspond to a first and a second region along the nanowires, such that etching takes place in the

second period at a higher current density than in the first period so that the nanowires formed have a greater diameter in the first region than in the second region.

The reference of Grüning discloses a method for etching cylindrical cavities with variable diameter along the axis (figure 2), electrochemical etching is used (column 3, line 45) and the etch diameter is controlled by the applied current density on the substrate which reads on the applicant's limitation of the anodic etching is carried out in a first time period and a second time period, which periods correspond to a first and a second region along the pores (nanowires).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the method of Van Den Meerakker by alternatively varying the current density to the substrate during etching to form a thinner diameter at the base of the nanowires for easy break-off at the end of the process because the reference of Grüning teaches how to etch a feature with variable diameter. One of ordinary skill would have been motivated to narrow the nanowires diameter in order to have an easy break-off point when applying mechanical stress such as ultrasonic vibration when separation of the nanowires from the substrate is desired.

A third difference is noted between applicants claim 1 and the reference of Van Den Meerakker cited above, Van Den Meerakker is silent about oxidizing a surface of the nanowires, whereupon said surface is removed by etching.

The reference of Bielgelsen teaches oxidation/etch steps are conventionally used for thinning quantum wires (column 7, line 65).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the method of Van Den Meerakker by adding oxidation steps for thinning the nanowires because Bielgelsen teaches oxidation/etch steps are conventionally used for thinning quantum wires. One of ordinary skill in the art would have been motivated to use oxidation in order to obtain thinner nanowires which is more desirable for higher integration requirements.

Claim Rejections - 35 USC § 103

4. Claims 2-5 are rejected under 35 U.S.C. 103(a) as being unpatentable over Van Den Meerakker et al. (Journal of The Electrochemical Society, 147 (7) pages 2757-2761 (2000)) in view of Grüning et al. (US 5,987,208), and Bielgelsen et al. (US 5,607,876) as applied to claim 1 above, and further in view of Kishi et al. (US 2003/0098640).

A difference is noted between applicants claim 2 and the reference of Van Den Meerakker cited above, Van Den Meerakker is silent about the removal taking place in a bath wherein a dispersion of nanowires is formed.

Kishi describes a method of separating nanotubes from a substrate citing the carbon nanotube can be extracted by applying ultrasonic wave to the carbon nanotubes dipped in a solvent such as ethanol so that they are peeled off the substrate (page 6, paragraph 0106) which reads on the applicant's limitation of forming a dispersion (page 6, paragraph 0112).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the method of Van Den Meerakker to include a

step for removing the nanowires in an ultrasonic bath wherein a dispersion is formed because the reference of Kishi teaches nanostructures such as tubes are conventionally removed from substrate using ultrasonic bath wherein a dispersion is formed. One of ordinary skill in the art would have been motivated to use an ultrasonic bath wherein a dispersion is formed in order to efficiently separate the nanowires from the substrate.

As to claim 3, It is noted that reference of Van Den Meerakker cited above is silent about the step of oxidation and removal is repeated.

However, if the nanowires length is shorter than the substrate thickness it would be obvious to one of ordinary skill in the art at the time the invention was made to further modify the method of Van Den Meerakker to use the same substrate to form more nanowires after the removing step because the structure needed for formation of new wires is left on the substrate after the previous removing step. One of ordinary skill in the art would have been motivated to use the same substrate to form more nanowires in order to save the cost of forming a new substrate.

As to claim 4, the limitation of "the anodic etching is carried out during a plurality of alternating first and second time periods so as to form a plurality of first and second regions which alternate along the lengths of the nanowires" has been addressed in the rejection of claim 1.

As to claim 5, the dispersion addressed in the rejection of claim 2 above will provide a layer of material and that material could be desired. Applicant has not shown any unexpected results related to the desirability of the material in claim 5.

Claim Rejections - 35 USC § 103

5. Claims 6-13 are rejected under 35 U.S.C. 103(a) as being unpatentable over Van Den Meerakker et al. (Journal of The Electrochemical Society, 147 (7) pages 2757-2761 (2000)) in view of Grüning et al. (US 5,987,208), and Bielgelsen et al. (US 5,607,876) and Kishi et al. (US 2003/0098640) as applied to claims 1-5 above, and further in view of Dennis et al. (US 2004/0076681).

A difference is noted between applicants claim 6 and the reference of Van Den Meerakker cited above, Van Den Meerakker is silent about a sol-gel process.

The reference of Dennis discloses "Silica nanotubes were prepared by the solgel method...This process yields silica nanotubes lining the pore walls plus silica surface films on both faces of the membrane" (page 10, paragraph 0121).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the method of Van Den Meerakker to include an additional step of providing a layer material by means of a sol-gel process because the reference of Dennis teaches sol-gel processes are conventionally used for lining nanostructures. One of ordinary skill in the art would have been motivated to use a sol-gel process to line the nanowires in order insure uniform coating around the wires, the sol-gel of Dennis provides silicon oxide (silica) but could also include a coloring agent. One of ordinary skill in the art would have found it obvious to include the coloring agent in the sol-gel if uniform coloration is preferred, rather than using a separate process for applying the coloring agent separately.

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As to claims 10, 11 It is noted that the reference of Van Den Meerakker is silent about wires length in a range of 0.3 to 1 μm .

The reference of Grüning discloses a an etch depth of 10 μ m in 10 minutes (column 5, line 47).

Therefore, it would have been obvious to one of ordinary skill in the art at the time the invention was made to modify the method of Van Den Meerakker to use the etch method of Grüning and adjust the time such that the wires length is in a range of 0.3 to 1 µm with an appropriate error margin. One of ordinary skill in the art would be motivated to adjust the etch time in order to obtain the desired etch depth within an appropriate error margin.

As to claim 12, 13 the modified method of Van Den Meerakker as described above would provide a method of manufacturing a device provided with nanowires on a substrate, in which method a dispersion of nanowires is provided on the substrate, characterized in that the dispersion as claimed in claim 7 is provided on the substrate. The dispersion can be used to manufacture any electronic device comprising a layer in which nanowires are dispersed, which nanowires have a predefined length distribution.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Mahmoud Dahimene whose telephone number is (571) 272-2410. The examiner can normally be reached on week days from 8:00 AM. to 5:00 PM..

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Nadine Norton can be reached on (571) 272-1465. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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MADINE NORTON:
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